# Alginate-crosslinked chitosan scaffolds as pentoxifylline delivery carriers

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Abstract To prevent fibrous encapsulation of implants, measures are taken to suppress inflammatory reactions around them. Sustained anti-inflammatory drug release from the scaffolds can potentially be a way to reduce inflammation around these implants. Alginate-crosslinked chitosan is often used to make biocompatible tissue engineered scaffolds. However, there is a lack of quantitative studies on the drug delivery properties of alginate-crosslinked chitosan scaffolds. For this study, chitosan, crosslinked with different concentrations of alginate, was made into porous scaffolds. Infrared and thermal gravimetric analyses showed polyelectrolyte complex formation between chitosan and alginate units. The alginate-crosslinked chitosan scaffolds were more hydrophilic, showed less swelling, had lower pentoxifylline (PTX) release efficacies, were more favorable for initial cell attachment, and were mechanically stronger and more resistant to enzymatic degradation when compared to non-crosslinked chitosan scaffolds. The differences became more significant as the concentrations of chitosan and alginate increased. Furthermore, in vitro tests showed that when PTX was slowly released from the scaffolds, it became more effective in suppressing the production of  $TNF-\alpha$  and IL-6 by stimulated macrophage cells.

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#### 1 Introduction

There has been an increasing interest in making biocompatible tissue engineered scaffolds with the capability to release drugs that promote successful tissue repair [\[1](#page-7-0)[–5](#page-8-0)]. Inflammation often arises from the insertion of foreign objects like implants into the body. This can cause fibrous/ collagenous encapsulation of the implants, which is the major cause for implant failures  $[6, 7]$  $[6, 7]$  $[6, 7]$  $[6, 7]$ . Anti-inflammatory medicines have been added to biopolymeric scaffolds to be released after surgical procedures for the purpose of suppressing inflammation caused by implant insertion [\[8–12](#page-8-0)]. It is believed that by suppressing inflammatory reactions after the insertion of an implant, the success rate of implant surgeries improves [\[13](#page-8-0)].

Pentoxifylline (PTX) is an anti-inflammatory drug that is known to prevent immune cells from producing inflammatory cytokines such as IL-1, IL-6 and TNF-a [\[14–16](#page-8-0)] and to treat or prevent tissue fibrosis [\[17–19](#page-8-0)]. PTX was reported to be able to inhibit phosphodiesterase, resulting in an increase in the intracellular levels of cAMP and a decrease in inflammatory cytokine production [\[14](#page-8-0), [15](#page-8-0)]. At a dosage of 100  $\mu$ g/mL, PTX can suppress immune cell proliferation [[16,](#page-8-0) [20](#page-8-0)]. Therefore, PTX was used as a model drug in this test.

Porous tissue-engineered scaffolds made of chitosanalginate polyelectrolyte complex have been reported with better biocompatibility  $[21-23]$  and drug release properties [\[24–28](#page-8-0)] than chitosan or alginate alone. Alginate is a brown seaweed extract that is regarded as safe by the FDA [\[29](#page-8-0)] and has been shown to have little or no immune reactions in the nearby tissue [\[30](#page-8-0), [31\]](#page-8-0). In vitro and in vivo results have shown that alginate is osteogenic [[21–23,](#page-8-0) [26](#page-8-0)]. Chitosan is a natural polymer that has been studied as a material for sustained drug release [[32–34\]](#page-8-0) and for tissue

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engineered scaffolds [\[35](#page-8-0), [36](#page-8-0)]. It is readily available, biocompatible, and bioresorbable. Alginate is often chemically crosslinked with chitosan to modify its abilities to uptake and release medicines [\[22–28](#page-8-0)].

Compounds such as anti-inflammatory and anti-bacterial drugs and growth factors have been added to implantable scaffolds to improve tissue repair [\[2](#page-7-0)[–5](#page-8-0), [8,](#page-8-0) [12](#page-8-0), [37\]](#page-8-0). However, their effects on cellular and tissue reactions were often evaluated qualitatively. The strength and concentrations of these compounds affect the consistency of their performance in vivo, which may lead to adverse side effects [\[38](#page-8-0)]. It is important to *quantitatively* characterize the scaffolds' drug release properties and how changes in these properties affect subsequent biological reactions.

In this study, chitosan and alginate-crosslinked chitosan porous scaffolds were tested as drug carriers. The effects that various degrees of crosslinking had on the pore structures, degradation rates, mechanical strengths, cell attachments, swelling ratios, and the PTX release efficacies of chitosan scaffolds were investigated. At the same time, the effects of PTXs release rates on its effectiveness in reducing inflammatory cytokine production by macrophage cells was also studied.

## 2 Materials and methods

All chemicals used in this study were purchased from Sigma, MO, USA, unless specified otherwise.

2.1 Preparation of chitosan and alginate-crosslinked chitosan solutions

Chitosan (96% deacetylation, C&B Co., Taiwan) was dissolved in a 2 v/v% acetic acid solution. Sodium alginate (Acros Organics, NJ, USA) was dissolved in deionized (DI) water. Chitosan and alginate solutions were stirred together for an hour before they were further processed into films and scaffolds. Five solutions were made. The final concentrations of the two polymers were 1% chitosan and no alginate (C1), 1% chitosan and 0.5% alginate (C1A0.5), 1% chitosan and 1% alginate (C1A1), 1% chitosan and 2% alginate (C1A2) and 2% chitosan and 1% alginate (C2A1). Films and porous scaffolds made from the five polymer solutions will be denoted as listed above. Films were used for contact angle and cell attachment tests and scaffolds were used for all other tests.

2.2 Making films and scaffolds

One milliliter of a polymer solution was coated on the bottom of each polystyrene petri dish  $(D = 3$  cm) and air dried at room temperature. Two milliliters of polymer solutions were poured into glass tubes and then frozen at  $-20^{\circ}$ C overnight. The frozen samples were then lyophilized in a freeze dryer (RVT4104, Savant, NY, USA). Films and lyophilized scaffolds were neutralized in 0.5 N NaOH for 30 min, rinsed with DI water three times, dehydrated using a series of ethanol solutions (20, 50, 70 and 100%) and air dried.

2.3 Thermo-gravimetric analysis (TGA)  $(n = 4)$ 

Dried scaffolds were cut into small pieces ( $\sim$ 10 mg) and put into a TGA instrument (Thermal analyst 2000, TA Instruments, USA). The TGA instrument's sample chamber was flushed with nitrogen gas before and during the test (60 mL/min). Scaffolds were heated from 25 to  $600^{\circ}$ C at  $10^{\circ}$ C/min and their weights were recorded over time. The percentage of derivative weight loss was plotted verses temperature to find the decomposition temperature.

2.4 Fourier transform infrared spectrum (FTIR)  $(n = 3)$ 

Scaffold specimens were ground, mixed with KBr, and pressed into tabs. They were then scanned from 4,000 to  $400 \text{ cm}^{-1}$  in a FTIR machine (Spectrum Gx, PerkinElmer, MA, USA) with a resolution of  $4 \text{ cm}^{-1}$ . Transmittance peaks were identified.

2.5 Contact angle  $(n = 4)$ 

Five microliters of DI water were placed on each polymer film. Static contact angles were measured using a contact angle meter (CA-D, Kyowa Interface Science Co., Japan).

2.6 Scaffold morphology by scanning electron microscope (SEM)  $(n = 3)$ 

Dried scaffolds were sliced and sputter coated with gold before being placed in the vacuum chamber of a SEM (Hitachi S-3000H, Japan). Microphotographs of the scaffold surfaces were taken.

2.7 Swelling ratio  $(n = 3)$ 

After measuring the diameters  $(D_0)$  of each dry scaffold ( $D \sim 1$  cm,  $h \sim 0.2$  cm), they were placed in phosphate buffered saline (PBS,  $pH = 7.4$ ) and their diameters were measured at each designated time point  $(D<sub>t</sub>)$ . The swelling ratio at each time point was calculated as  $[(D_t - D_0)$  $D_{\rm o}$   $\times$  100%.

#### 2.8 Release efficiency of pentoxifylline (PTX)  $(n = 3)$

Each dry scaffold disc ( $D \sim 1$  cm,  $h \sim 0.2$  cm) was loaded with  $100 \mu L$  PTX (20 mg/mL) and left to dry at room temperature overnight. Ten scaffold discs were placed in a beaker filled with 100 mL of PBS, three beakers  $(n = 3)$  for each polymer. All beakers were placed on a shaker (50 rpm) during the test. At each designated time point,  $100 \mu L$  of PBS was sampled and the same volume of PBS was then replenished. The absorbance of PTX at 274 nm was measured (Gene Quant 1300 spectrophotometer, GE Healthcare, NJ) and converted to concentrations  $(M_t \text{ (µg/mL)})$  using a standard curve. The release efficiency of the PTX was calculated as  $[(M_t/200 \text{ µg/mL})] \times 100\%$ . After the PTX in the ten scaffolds was released, the final PTX concentration in each beaker was 200 µg/mL.

#### 2.9 Degradation ( $n = 3$ )

The chitosan used to make samples for the degradation test was 82% deacetylated because it is more easily degraded by lysozyme than 96% deacetylated chitosan [[39\]](#page-8-0). A scaffold with an initial dry weight of  $W_0$  was placed in a 15 mL centrifuge tube filled with 5 mL of lysozyme solution (70,000 U/mL PBS,  $pH = 7.4$ ). The tubes were placed in an incubator  $(37^{\circ}C)$  and the lysozyme solution was changed every 3 days. On days 7, 14 and 21, three tubes of each specimen were removed from the incubator, and the remaining scaffolds were rinsed carefully several times with DI water and air dried to obtain their dry weights (W). The percentage of weight loss was calculated as  $[(W_0 - W)/W_0] \times 100\%$ .

#### 2.10 Compression  $(n = 4)$

Dry scaffolds ( $D \sim 1$  cm,  $h \sim 1.2$  cm) were submerged in PBS overnight and excess liquid was removed before testing. A Universal Micro-tribometer (model UMT-2, CETR, CA, USA) was used to obtain stress–strain curves. The exact diameter and height of each scaffold was measured right before the test. A 10 N load cell was used and the speed of the crosshead was set at 0.3 mm/s. The test ended when the crosshead reached 80% strain of the scaffold. The slope of the initial linear section of the stress– strain curve was used to estimate the Young's modulus.

#### 2.11 Cell attachment  $(n = 4)$

The bottom of each well of a 24-well tissue culture polystyrene (TCP) plate (BD Biosciences, CA) was coated with 200 μL of polymer solution and air dried. The TCP surface served as positive control. Films were processed as described earlier. Coated films were disinfected with 70% ethanol solution, exposed to UV briefly, and then rinsed with sterile PBS to remove residual ethanol. Osteoblasts (7F2, ATCC, VA)  $(3 \times 10^5 \text{ cells/mL})$  were seeded onto the films  $(1 \text{ mL/well})$ . Cells were cultured in  $\alpha$ -MEM (Invitrogen, CA, USA) without serum. The culture was maintained in a cell culture incubator (SCA-165DS, ASTEC, Japan) at 95% humidity, 37 $\degree$ C, and 5% CO<sub>2</sub>. At designated time points, unattached cells were counted to back calculate the number of attached cells.

2.12 In vitro test of anti-inflammatory effect during controlled release  $(n = 4)$ 

The test was done using C1 and C1A2 scaffolds to compare the effects caused by varying degrees of slow release. Scaffolds were each loaded with 50  $\mu$ L of PTX (4 mg/mL) and allowed to dry at room temperature. Macrophage cells (RAW 264.7, ATCC, VA) were seeded into 24-well TCP plates at  $1 \times 10^5/\text{cm}^2$  and allowed to settle overnight in a cell culture incubator. Cell culture media (DMEM plus 10% fetal bovine serum, 100 U/mL penicillin, 100 μg/mL streptomycin) was then removed, and the cells were separated into four different test groups—(1) DMEM: cells cultured in DMEM with a blank scaffold,  $(2)$  DMEM + LPS: cells in DMEM  $+ 1 \mu g/mL$  LPS  $+$  blank scaffold, (3) DMEM + LPS + PTX (in media): cells in DMEM + 1 µg/mL LPS  $+$  100 µg/mL PTX  $+$  blank scaffold, and (4) DMEM  $+$  LPS  $+$  PTX (from scaffold): cells in DMEM  $+ 1 \mu$ g/mL LPS  $+$  scaffold loaded with 100  $\mu$ g PTX [[16\]](#page-8-0). Each well contained 2 mL of medium.

Lipopolysaccharide (LPS,  $1 \mu g/mL$ ) was used to activate the macrophage cells to release inflammatory factors [\[40\]](#page-8-0). Six hours later, tumor necrosis factor alpha (TNF- $\alpha$ ) and interleukin 6 (IL-6) in the media were measured using commercial kits (Mouse TNF-alpha ELISA OptiEIA, BD Bioscience, CA, and Mouse IL-6 ELISA DuoSet, R&D, MN).

#### 2.13 Statistical analysis

All data is presented as average  $\pm$  standard deviation. Statistical analysis was conducted using two-way ANOVA with Tukey tests (SigmaStat 3.5 for Windows, Systat Software, IL). Difference between alginate-crosslinked chitosan and chitosan was declared when  $P < 0.05$ .

## 3 Results and discussion

## 3.1 TGA and FT-IR

Both TGA and FTIR were used to characterize bond formation between chitosan and alginate. The decomposition

<span id="page-3-0"></span>temperatures of all polymer samples are listed in Table 1. Compared to chitosan  $(306^{\circ}C)$ , the decomposition temperatures of alginate-crosslinked chitosan decreased as the concentrations of alginate increased  $(P<0.05)$ . The increase in chitosan concentration had no significant effect on the temperature since C2A1 had a similar decomposition temperature as C1A1 ( $P > 0.05$ ). The decomposition temperature of pure alginate was  $251.3^{\circ}$ C. The decomposition temperatures of alginate-crosslinked chitosan fell between those of chitosan and alginate, indicating that covalent bond formation did not take place [[41,](#page-8-0) [42](#page-8-0)]. Alginate, an anionic polysaccharide, will form polyelectrolyte complex with chitosan, a cationic polysaccharide, through ionic attraction between the  $NH_3^+$  of chitosan and the  $COO<sup>-</sup>$  of alginate  $[22, 23, 43]$  $[22, 23, 43]$  $[22, 23, 43]$  $[22, 23, 43]$  $[22, 23, 43]$  $[22, 23, 43]$  $[22, 23, 43]$ .

The carboxylate vibration bands of alginate are typically at 1,412 cm<sup>-1</sup> (symmetric) and 1,596 cm<sup>-1</sup> (asymmetric vibration) [\[44](#page-8-0)]. Chitosan's amine group has typical absorbance at 1,587 and 1,647 cm<sup>-1</sup> (amide I). The 1,587 cm<sup>-1</sup> peak is assigned to the N–H bending vibration of amine overlapping amide II vibration [[44\]](#page-8-0). After chitosan was crosslinked with alginate, three new peaks emerged at 1407, 1554, and 1616  $cm^{-1}$  (Fig. 1). The XPS analysis from Lawrie et al.  $[44]$  $[44]$  showed that the 1,616 cm<sup>-1</sup> peak was due to amide I and asymmetric  $NH_3^+$  deformation. Peak at 1,554  $cm^{-1}$  was assigned to amide II (9%) overlapping amine (36%) and protonated amine (55%). The intensity of the peak at  $1,554$  cm<sup>-1</sup> decreased as the amount of alginate increased. The peak at  $1,407$  cm<sup>-1</sup> was assigned to the alginate's carboxylate vibration.

#### 3.2 Contact angle

The contact angles (Table 1) of alginate-crosslinked chitosan were all smaller than that of chitosan ( $P < 0.05$ ) except for C2A1). As the ratio of alginate increased, the contact angles of crosslinked chitosan got smaller. C2A1 had a higher contact angle compared to that of C1A1  $(P<0.05)$ .



Fig. 1 Fourier transform infrared (FTIR) spectra of chitosan and alginate-crosslinked chitosan scaffolds

Alginate is water soluble and more hydrophilic than chitosan. As the concentration of chitosan increases in the chitosan-alginate complex, the hydrophilicity decreases. On the other hand, as the percentage of alginate increases in the complex, its hydrophilicity increases.

## 3.3 SEM

As more alginate was added to chitosan, the pores of the chitosan scaffolds became more closed (less interconnectivity) and smaller. The texture of the scaffold walls became flaky as the percentage of alginate increased (Fig. [2\)](#page-4-0). Increasing chitosan concentration (Fig. [2](#page-4-0)e) resulted in more honeycomb-like pores and a less flaky texture.

Chung et al. [[23](#page-8-0)] reported decreased porosity and pore size after chitosan was crosslinked with alginate. The scaffolds had fewer pores and a more closed structure. Polyelectrolyte complex and ionic bonds between the two polymers were thought to result in a less porous structure [\[23](#page-8-0), [45](#page-9-0)].

Table 1 Decomposition temperatures, contact angles, swelling ratios and Young's moduli of chitosan and alginate-crosslinked-chitosan films and scaffolds

	1. Decomposition temperature $({}^{\circ}C)^*$	2. Contact angle $(°)$	3. Swelling ratio at 90 min $(\%)$	4. Young's modulus (kPa)
C1	$306.3 \pm 0.3^{\circ}$	$82.8 \pm 3.1^{\circ}$	$34.3 \pm 1.1^{\circ}$	$2.3 \pm 0.2^{\rm a}$
C1A0.5	$299.5 \pm 0.7^{\rm b}$	$69.0 \pm 2.9^b$	$32.8 \pm 0.2^b$	$3.1 \pm 0.0^b$
C1A1	$269.4 \pm 0.7^{\circ}$	$55.5 \pm 2.6^{\circ}$	$29.5 \pm 0.1^{\circ}$	$3.5 \pm 0.4^{\circ}$
C1A2	$266.4 \pm 0.7$ <sup>d</sup>	$48.0 \pm 1.6^{\rm d}$	$27.8 \pm 0.2^d$	$6.8 \pm 0.0^d$
C2A1	$298.1 \pm 1.5^{\rm b}$	$77.0 \pm 2.4^{\rm a}$	$23.9 \pm 0.1^e$	$9.6 \pm 0.9^e$

Data represent the mean  $\pm$  standard deviation ( $n = 4$ , two-way ANOVA). Numbers marked with different letters are statistically different (only for numbers within the same column)

\* The onset decomposition temperature of pure alginate is  $251.3 \pm 1.4^{\circ}$ C

<span id="page-4-0"></span>Fig. 2 Representative SEM microphotographs of a C1 (1% chitosan), b C1A0.5 (1% chitosan  $+$  0.5% alginate), c C1A1 (1% chitosan + 1%) alginate), d C1A2 (1%  $chitosan + 2\%$  alginate), e C2A1 (2% chitosan + 1%) alginate) scaffolds. Scale  $bar = 500 \mu m$  for all micrographs



 $(e)$  C<sub>2A1</sub>

## 3.4 Swelling ratio

Figure [3](#page-5-0) shows the swelling ratios of the scaffolds during the first 60 min of the test. The swelling ratios of the scaffolds after 90 min in PBS are listed in Table [1.](#page-3-0) Scaffolds crosslinked with alginate swelled up more slowly than uncrosslinked chitosan scaffolds. The higher the alginate concentration was, the slower the scaffold swelled. The swelling ratios of alginate-crosslinked scaffolds were significantly smaller than those made of uncrosslinked chitosan ( $P < 0.05$ ). Increased concentrations of chitosan also reduced the scaffold's swelling ratio  $(P<0.0.5)$ .

The chains of the two polymers formed a branched network through electrostatic attraction between the two oppositely charged polymer chains. The branched network prevents polymer chains from stretching and swelling freely [[22\]](#page-8-0). Without the ionic bonding, scaffolds with higher polymer mass would have swelled more at the end of the test. Compared with C1A1, the increase in chitosan percentage in C2A1 made its scaffolds more hydrophobic and had smaller pores. As a result, the C2A1 scaffolds swelled less than the C1A1 scaffolds.

#### 3.5 Release efficiency

By 90 min, C1, C1A0.5, C1A1, C1A2, and C2A1 scaffolds released 93.4, 85.7, 83.1, 78.6, and 70.7% of their total PTX respectively (Fig. [4\)](#page-5-0). As the percentages of alginate in the scaffolds increased, the scaffolds' release efficiencies

<span id="page-5-0"></span>

decreased. C2A1 had the lowest release efficacy and lowest initial PTX burst of all samples.

The scaffolds made from solutions with higher polymer concentrations had higher polymer densities, which resulted in lower mass transportation inside the scaffolds. Vandenberg et al. [[27\]](#page-8-0) observed lower protein release efficiency in alginate microcapsules coated with higher concentrations of chitosan. Between groups with similar polymer mass (C2A1 and C1A2), C2A1 released less PTX over time. Besides having a smaller swelling ratio, C2A1 was more hydrophobic and had smaller pores, which slowed down aqueous-phase mass transfer. Chitosan microspheres crosslinked with alginate were shown to have less release capability for lipophilic drugs [[24\]](#page-8-0).

#### 3.6 Degradation

The addition of alginate significantly reduced the percentage of degradation in the chitosan scaffolds compared to the uncrosslinked chitosan scaffolds (Fig. [5](#page-6-0)). Takeuchi et al. [\[28](#page-8-0)] observed a lower eroding rate in chitosan-alginate composite particles than in alginate particles. Between scaffolds with similar polymer mass, i.e. C1A1 and C1A2, their percentages of degradation were similar throughout the test. The percentages of degradation seemed to be more dependent on polymer density than the degree of crosslinking.

Lysozyme targets and hydrolyzes glycosidic bonds (C–O–C) [[46–48\]](#page-9-0) of chitosan and alginate polymers. Ionic attraction due to crosslinking would have minimal effect on preventing lysozyme from interacting with the polymer complex.

#### 3.7 Compression

The Young's moduli of chitosan scaffolds increased as the concentrations of alginate and chitosan increased  $(P<0.05)$  (Table [1](#page-3-0)). Compared to C1, the moduli of



Fig. 4 The release efficacies of chitosan and alginatecrosslinked chitosan scaffolds

<span id="page-6-0"></span>

Fig. 5 The percentages of weight loss of chitosan and alginatecrosslinked chitosan scaffolds due to enzymatic degradation over time

C1A0.5 and C1A1 were not significantly higher. However, the modulus of the C1A2 scaffold was almost twice as much as that of C1A1. The higher polymer mass and smaller pores may have resulted in the higher compressional Young's moduli of the C1A2 scaffolds. With a similar amount of polymer mass, C1A2 had a lower modulus than C2A1. Hence besides polymer mass, different degrees of crosslinking could also cause differences in scaffolds' moduli.

Li et al. [\[22](#page-8-0)] reported chitosan and alginate-crosslinked chitosan scaffolds to have Young's moduli of 2.56 and 8.16 MPa respectively. They had used a 4.8% chitosanalginate 1:1 mixture instead of the 2% in this test and their test was done on dry scaffolds. Nonetheless, we can see

Fig. 6 Percentages of initial cell attachment on chitosan and alginate-crosslinked chitosan surfaces

that alginate crosslinking has significantly increased the compressional moduli of chitosan scaffolds.

#### 3.8 Cell attachment

After 1 and 1.5 h, films with higher alginate content were more favorable for cell attachment (Fig. 6). At 2 h, C2A1 had the lowest percentage of cell attachment of all  $(P<0.05)$ .

Most cells prefer to attach to more hydrophilic surfaces [\[49](#page-9-0)]. Increased cell attachment and proliferation on chitosan scaffolds after they had been crosslinked with alginate have also been seen by others [\[21](#page-8-0), [22\]](#page-8-0). C2A1 and C1 had similar hydrophobicity (Table [1](#page-3-0)) and similar percentages of initial cell attachment.

## 3.9 Anti-inflammatory effect of PTX slow release

There was a significant difference between the release efficiencies of C1 and C1A2, hence the test was done on the two specimens. Macrophage cells did not release TNF- $\alpha$  and IL-6 in the presence of scaffolds (DMEM groups in Fig. [7](#page-7-0)a, b) over the first 24 h. With  $1 \mu g/mL$  of LPS  $(DMEM + LPS)$ , macrophage cells released similar amounts of TNF- $\alpha$  and IL-6 between the two groups. When PTX was present in the media containing LPS  $(DMEM + LPS + PTX$  (in media)), the amounts of TNF- $\alpha$  and IL-6 were slightly lower than those in the  $DMEM + LPS$  group. There was no significant difference between the two groups since PTX did not come from the scaffolds. When scaffolds were loaded with PTX, and PTX was slowly released it into media containing LPS



<span id="page-7-0"></span>

Fig. 7 Tumor necrotic factor alpha (TNF- $\alpha$ ) (a) and interleukin 6 (IL-6) (b) release from the four conditions: DMEM: cells cultured in DMEM with a blank scaffold,  $DMEM + LPS$ : cells in  $DMEM +$ 1  $\mu$ g/mL LPS + blank scaffold, DMEM + LPS + PTX (in media): cells in DMEM  $+ 1 \mu g/mL$  LPS  $+ 100 \mu g/mL$  PTX  $+$  blank scaffold, DMEM  $+$  LPS  $+$  PTX (from scaffold): cells in DMEM  $+$ 1 µg/mL LPS + scaffold loaded with 100 µg PTX

 $(DMEM + LPS + PTX$  (from scaffold)), the amounts of TNF- $\alpha$  and IL-6 were significantly lower than those in  $DMEM + LPS + PTX$  (in media) ( $P \lt 0.05$ ). Furthermore, the amounts of IL-6 and TNF- $\alpha$  were significantly lower in C1A2 than in C1 ( $P < 0.05$ ).

The slow release of PTX more effectively prevented macrophages from releasing pro-inflammatory factors compared to the direct delivery of PTX into the media. PTX was found to be able to decrease oxidative stress during inflammation and to suppress the superoxide production of macrophage [\[14](#page-8-0), [17\]](#page-8-0). Based on the three possible cellular metabolic pathways of PTX  $[15]$  $[15]$ , it is speculated that in this test when a high dosage of PTX was added to macrophage cells, it was used not only in suppressing TNF- $\alpha$  mRNA and the sequential IL-6 expressions, but also in inhibiting cell proliferation [[20\]](#page-8-0). In other words, PTX was metabolized via several pathways. However, when PTX was slowly released from a scaffold, its concentration went from zero to an optimal concentration  $(\leq 10 \text{ µg/mL}$  [\[20](#page-8-0)]) that could suppress TNF- $\alpha$  expression but not cell growth or superoxide production. The slow release thus allowed for a more efficient use of PTX in macrophages in reducing the expression of TNF-a. Kofuji et al. [\[50](#page-9-0)] implanted chitosan beads retaining the antiinflammatory drug prednisolone (PS) into the back of mice and investigated the therapeutic efficacy of sustained PS release against local inflammation. They found that the implantation of chitosan beads loaded with PS significantly reduced inflammation compared with the injection of PS suspension. They concluded that sustained PS release provides a minimum effective dose and facilitated prolonged periods of local drug presence. With the direct injection of PS suspension, most of the drug would be removed in the process of diffusion because of its high concentration gradient.

## 4 Summary

The concentration of alginate significantly changed the properties of the chitosan scaffolds measured in this test. It was hard to discern whether the changes in chitosan scaffolds' swelling ratio, release efficacy, degradation, and compressive Young's modulus were the results of alginate crosslinking or of increases in polymer mass in the scaffolds. Our results also show that a slow release of PTX can more effectively suppress inflammatory reactions in macrophage cells.

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